

Fragmentation of multiply-charged small hydrocarbon molecules C_nH^{q+} ($n=1-3, q=2-6$) produced in high velocity collisions: Branching ratios and kinetic energy releases of the H^+ fragment

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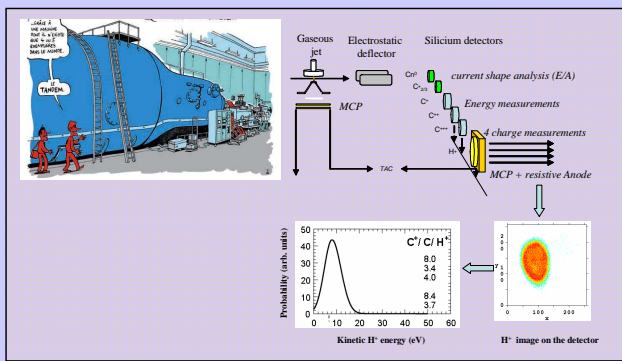
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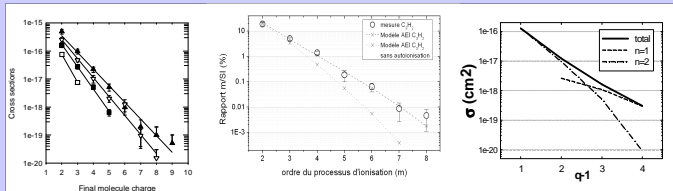
In a recent work (Chabot et al this conference and PRL **104**, 043401 (2010)), we showed how fragmentation of multi-charged carbon clusters C_n^{q+} strongly evolves with the strength of the coulomb interaction. In this work, we pursued these studies on new C_nH^{q+} species ($n=1-3, q=2-6$) and added a new experimental observable: the kinetic energy release of the H^+ fragment $KER-H^+$, extracted for each dissociation channel.

Experimental method

AGAT set-up at the Tandem accelerator in Orsay



Ionization cross sections

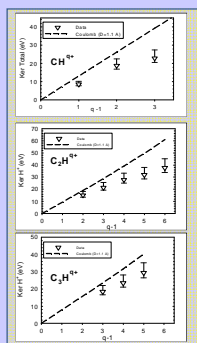


Measured ionization cross sections of in C_nH^{q+} -He collisions at $v=4.5$ a.u.; lines are to guide the eye (left). Comparison between measured and calculated (IAE model) ionization cross sections in $C_3H_2^+$ at $v=4.5$ a.u. (middle). Calculated (IAE model) valence and 1s ionization cross sections in CH^+ -He at $v=3.6$ a.u. (right)

KER- H^+ measurements

& comparison with the point charge coulomb model

Mol.	Charge	Channel	E1c (eV)	σ_1 (eV)
CH	1	CH ⁺	29.3	3.0
		C ⁺ H	7.9	2.7
		C ²⁺ H	16.4	6.5
		C ³⁺ H	20.7	8.4
C ₂ H	1	C ₂ H ⁺	0.9	2.8
		CC ⁺ H	1.2	3.1
		C ₂ H ²⁺	6.0	2.0
		C ⁺ C ⁺ H	8.4	3.7
		C ²⁺ C ⁺ H	15.1	4.7
		C ³⁺ C ⁺ H	18.1	6.6
C ₃ H	1	C ₃ H ⁺	21.4	7.6
		C ²⁺ C ₂ H	11.4	13.1
		C ³⁺ C ⁺ H	27.8	9.3
		C ⁴⁺ C ⁺ H	24.5	10.3
		C ⁵⁺ C ⁺ H	21.8	10.8
		C ⁶⁺ C ⁺ H	17.7	7.1
C ₂ H	2	C ₂ H ²⁺	1.5	1.4
		C ²⁺ C ⁺ H	1.3	2.1
		CC ²⁺ H	6.5	6.6
		C ₂ H ³⁺	8.5	4.2
		C ³⁺ C ⁺ H	6.1	2.7
		C ⁴⁺ C ⁺ H	5.3	2.5
C ₃ H	3	C ₃ H ³⁺	7.3	3.3
		C ₂ ⁺ C ⁺ H	10.7	3.5
		C ³⁺ C ²⁺ H	12.7	3.8
		C ⁴⁺ C ⁺ H	9.7	6.2
		C ⁵⁺ C ⁺ H	12.9	4.2
		C ⁶⁺ C ⁺ H	18.4	6.6
C ₃ H	4	C ₃ H ⁴⁺	13.1	8.1
		CC ³⁺ H	17.3	8.5
		C ⁴⁺ C ₂ H	23.4	9.1
		C ⁵⁺ C ⁺ H	28.0	4.4
		C ⁶⁺ C ⁺ H	29.4	9.5
		C ⁷⁺ C ⁺ H	26.1	11



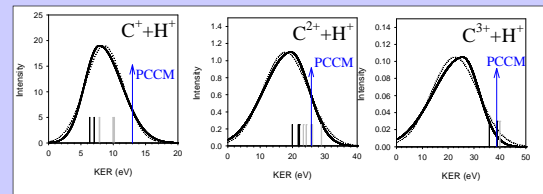
Characteristics of measured $KER-H^+$ distributions for various dissociation channels of C_nH^{q+} molecules. A Gaussian shape (E1c: peak position, σ_1 : standard deviation was assumed) (left)

Comparison between measured KER values and predictions of the point charge coulomb model (PCCM) (right)

KER calculations (CH^{q+} case)

State	CH^+	CH^{2+}	CH^{3+}	CH^{4+}
Ground state	$X^1\Sigma^+(0)$ ($1\sigma^2 2\sigma^2 3\sigma^2$)	$1^1\Sigma^+(0)^{**}$ ($1\sigma^2 2\sigma^2 3\sigma$)	$1^1\Sigma^+(0)^{**}$ ($1\sigma^2 2\sigma^2$)	$1^2\Sigma^+(0)^{**}$ ($1\sigma^2 2\sigma$)
First Excited state	$a^1\Pi(0.98)$ ($1\sigma^2 2\sigma^2 3\sigma 1\pi$)	$1^3\Pi(3.6)^{**}$ ($1\sigma^2 2\sigma^2 1\pi$)	$1^3\Sigma^+(5.9)^{**}$ ($1\sigma^2 2\sigma 3\sigma$)	$2^2\Sigma^+(11)^{**}$ ($1\sigma^2 3\sigma$)
Second excited state	$2^2\Sigma^+(10)^*$ ($1\sigma^2 2\sigma 3\sigma^2$)	$1^3\Pi(8.4)^{**}$ ($1\sigma^2 2\sigma 3\sigma$)	$1^3\Pi(12.1)^{**}$ ($1\sigma^2 1\pi$)	
Third excited state	$2^3\Pi(13.1)^{**}$ ($1\sigma^2 2\sigma 3\sigma 1\pi$)	$2^3\Sigma^+(10.1)^{**}$ ($1\sigma^2 2\sigma 3\sigma$)	$3^2\Sigma^+(31.7)$ ($1\sigma^2 4\sigma$)	
Fourth excited state	$3^3\Pi(15.2)^{**}$ ($1\sigma^2 2\sigma 3\sigma 1\pi$)	$1^3\Pi(13.5)^{**}$ ($1\sigma^2 2\sigma 1\pi$)		
Fifth excited state	$1^3\Delta(16)$ ($1\sigma^2 2\sigma 1\pi^2$)	$2^3\Pi(20.5)^{**}$ ($1\sigma^2 3\sigma 1\pi$)		
Sixth excited state	$3^3\Sigma^+(17.5)$ ($1\sigma^2 2\sigma^2 4\sigma$)	$3^1\Sigma^+(21.1)^*$ ($1\sigma^2 3\sigma^2$)		
Seventh excited state	$1^4\Sigma^+(18.2)$ ($1\sigma^2 2\sigma 1\pi^2$)	$1^4\Sigma^+(22.85)$ ($1\sigma^2 1\pi^2$)		
Eighth excited state		$2^3\Pi(24.3)^{**}$ ($1\sigma^2 3\sigma 1\pi$)		

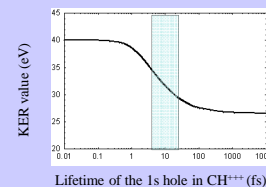
Calculated electronic excited states of CH^+ , CH^{2+} , CH^{3+} and CH^{4+} by the CASSCF method using the MOLPRO package



Comparison between measured and calculated $KER-H^+$

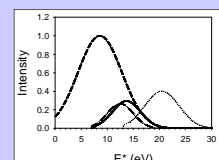
Dynamics associated to Auger relaxation

1s ionization contributes to 80% to the CH^{4+} production cross section. The dynamics of relaxation of the 1s hole, of the order of 10fs in $C^{(4+/+)}$, strongly influences the KER value, as calculated within the PCCM model (figure below)



Partitioning of energy in (C_3H)⁺⁺

Partitioning of energy in the fragmentation of $(C_3H)^{2+}$ amongst 4 channels: C_3^+/H^+ (short dashed line), $C_2^+/C^+/H^+$ (dash-dot line), $C_2^+/C^2+/H^+$ (solid line) and $C/C^2+/H^+$ (dotted line).



$E^* = E_{\text{formation}} + KER-H^+$