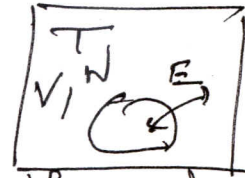


I)

1)  $E$  fluctue

1)  $T = \text{cte}$ ,  $V = \text{cte}$ ,  $N = \text{cte}$



2) tout système en équilibre thermique avec un thermostat à la température  $T$  la contribution à l'énergie moyenne de chaque variable intervenant au carré et uniquement au carré, dans l'expression de l'énergie, vaut

1)  $\frac{1}{2} k_B T$

$$E = ap^2 + bq^2$$

$\nearrow \frac{k_B T}{2}$      $\nearrow \frac{k_B T}{2}$

3)  $\langle E \rangle = E = E_{\text{trans}} + E_{\text{rot}} = H_{\text{trans}} + H_{\text{rot}}$

1)  $E = \frac{p_x^2}{2m} + \frac{p_y^2}{2m} + \frac{p_z^2}{2m} + \frac{p_\theta^2}{2I} + \frac{p_\phi^2}{2I \sin^2 \theta}$

$\langle E \rangle = \frac{5}{2} k_B T$  par 1 particule

pour les  $N$  molécules  $\langle E \rangle = N \langle E \rangle = \frac{5}{2} N k_B T$

1)  $C_V = \left( \frac{\partial \langle E \rangle}{\partial T} \right)_{V, N} = \frac{5}{2} N k_B$

$C_p - C_V = N k_B \Rightarrow C_p = \frac{7}{2} N k_B$

1)  $\gamma = \frac{C_p}{C_V} = \frac{7}{5} = 1,4$

II

$$1) \int \rho = e^{-\beta \epsilon_1} + e^{-\beta \epsilon_2} = 1 + e^{-\beta \epsilon}$$

$$Z = \rho^N = (1 + e^{-\beta \epsilon})^N$$

$$2) P_1 = \frac{1}{\rho} e^{-\beta \epsilon_1} = \frac{1}{1 + e^{-\beta \epsilon}}, \quad P_2 = \frac{1}{\rho} e^{-\beta \epsilon_2} = \frac{e^{-\beta \epsilon}}{1 + e^{-\beta \epsilon}}$$

$$P_1 + P_2 = 1$$

$$P_2 = \frac{1}{e^{\beta \epsilon} + 1}$$

$$3) N_1 = P_1 N = \frac{N}{1 + e^{-\beta \epsilon}}$$

$$N_2 = P_2 N = \frac{N}{e^{\beta \epsilon} + 1}$$

$$N_1 + N_2 = N$$

$$4) \theta \neq 0 \quad \frac{\epsilon}{k_B T} = \beta \epsilon = \frac{\theta}{T}$$

$$5) T \ll \theta \quad \text{Basses Températures} \quad N_1 = \frac{N}{1 + e^{-\frac{\theta}{T}}} \quad N_2 = \frac{N}{e^{\frac{\theta}{T}} + 1}$$

$$N_1 \rightarrow N$$

$$N_2 \rightarrow N e^{-\theta/T} \rightarrow 0$$

→ tous les molécules sont au niveau le + bas

$T \gg \theta$  Hautes températures

$$N_1 \rightarrow \frac{N}{1 + 1 - \frac{\theta}{T}} \sim \frac{N}{2}$$

$$\frac{\epsilon}{k_B T} \ll 1 \quad \epsilon \ll k_B T$$

$$N_2 \rightarrow \frac{N}{2 + \frac{\theta}{T}} \sim \frac{N}{2}$$

les molécules se répartissent de manière égale sur les 2 niveaux

écart entre les 2 niveaux est négligeable

$$6) \langle E \rangle = 1 N_1 \epsilon_1 + N_2 \epsilon_2 = \frac{N \epsilon}{e^{\beta \epsilon} + 1}$$

$$\langle E \rangle = - \frac{\partial}{\partial \beta} (\ln Z) = -N \frac{\partial}{\partial \beta} (\ln \rho) = N \frac{\epsilon e^{-\beta \epsilon}}{1 + e^{-\beta \epsilon}} = \frac{N \epsilon}{e^{\beta \epsilon} + 1}$$

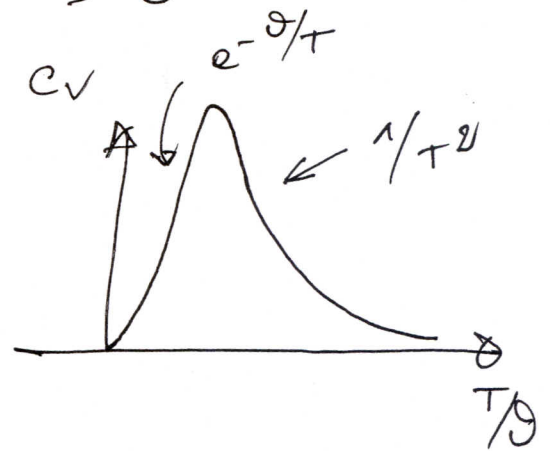
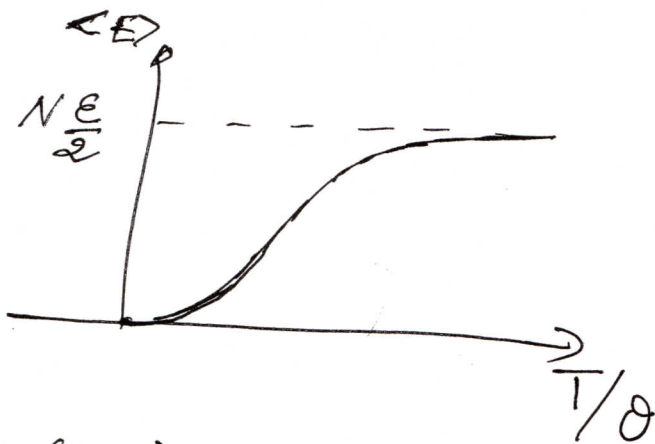
$$\langle E \rangle = \frac{NE}{e^{\frac{\theta}{T}} + 1} = Nk_B \frac{\theta}{e^{\frac{\theta}{T}} + 1}$$

$\frac{T}{\theta} \gg 1$        $T \gg \theta$       High Temperature

$$\langle E \rangle \rightarrow \frac{NE}{2 + \frac{\theta}{T}} \sim \frac{NE}{2}$$

$\frac{T}{\theta} \ll 1$        $T \ll \theta$       Low Temperature

$$\langle E \rangle \rightarrow \frac{NE}{e^{\frac{\theta}{T}} + 1} \rightarrow 0$$



$$7) C_V = \left( \frac{\partial \langle E \rangle}{\partial T} \right)_{V,N} = \left( \frac{\partial \langle E \rangle}{\partial \beta} \right)_{V,N} \left( \frac{\partial \beta}{\partial T} \right)_{V,N} = -\frac{1}{k_B T^2} \left( \frac{\partial \langle E \rangle}{\partial \beta} \right)_{V,N}$$

$$C_V = -\frac{k_B N}{(k_B T)^2} \cdot \left( \frac{-N e^2 e^{\beta E}}{(e^{\beta E} + 1)^2} \right) = \frac{N k_B}{(k_B T)^2} \frac{e^2 e^{\beta E}}{(1 + e^{\beta E})^2}$$

$$C_V = N k_B \left( \frac{E}{k_B} \right)^2 \frac{1}{T^2} \frac{e^{\beta E}}{(1 + e^{\beta E})^2}$$

$$C_V = N k_B \left( \frac{\theta}{T} \right)^2 \frac{e^{\beta E}}{(1 + e^{\beta E})^2} = N k_B \left( \frac{\theta}{T} \right)^2 \frac{e^{+\theta/T}}{(1 + e^{\theta/T})^2}$$

$T \rightarrow 0$       ( $T \ll \theta$ )       $C_V \rightarrow N k_B \left( \frac{\theta}{T} \right)^2 e^{-\theta/T} \rightarrow 0$

$T \rightarrow \infty$       ( $T \gg \theta$ )       $C_V \rightarrow N k_B \left( \frac{\theta}{T} \right)^2 \frac{(1 + \theta/T)}{(2 + \theta/T)^2} \rightarrow 0$

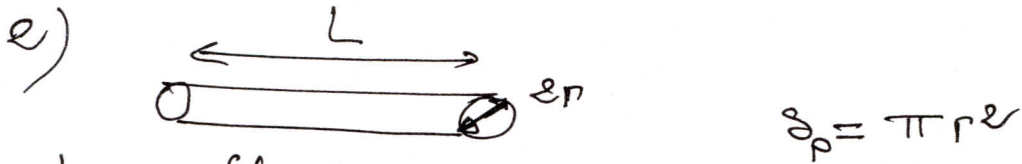
$C_V \rightarrow \frac{N k_B}{1} \left( \frac{\theta}{T} \right)^2 \sim d \frac{1}{T^2} \rightarrow 0$

III

1)  $\vec{J}_n = -D \underbrace{\text{grad } n_V}_{m^{-4}} = -D \frac{\partial n_V}{\partial x} \vec{e}_x$   $\rightarrow$  densité volumique ( $m^{-3}$ )

coefficient de diffusion ( $\frac{s^{-1} m^{-2}}{m^{-4}} \equiv s^{-1} m^2$ )

vecteur normal volumique de partants ( $s^{-1} m^2$ )



$$\Phi_p = \iint_S \vec{J}_n \cdot \vec{dS} = \iint_S \vec{J}_n \cdot \vec{n} dS$$

$$\Phi_p = \iint_S -D \frac{\partial n_V}{\partial x} (\vec{e}_x \cdot \vec{n}) dS$$

$$\vec{n} \equiv -\vec{e}_x \quad \Phi_p = +D \frac{dn_V}{dx} \iint_S dS$$

$$\Phi_p = D \frac{dn_V}{dx} \pi r^2$$

$$\frac{dn_V}{dx} \approx \frac{n_{2V}(L) - n_{1V}(L)}{e} = -\frac{\Delta(T)}{e}$$

$$\Phi_p \approx -\frac{D \Delta(T)}{e} \pi r^2 \Rightarrow$$

3)  $\Phi_m = \rho \times S \times \Phi_p = -\frac{D \pi r^2 \rho S}{e} \Delta(T)$

$$= -K \Delta(T) S$$

$$\text{avec } K = \frac{\rho \pi r^2 D}{e}$$

$$\Rightarrow r = \sqrt{\frac{eK}{\rho \pi D}} = \sqrt{\frac{10^{-5} \times 10^{-6}}{10^{10} \times \pi \times 10^{-9}}} = 0,56 \mu m$$

$$h) \quad -\frac{dN_1(t)}{dt} = \frac{dN_2(t)}{dt} = \phi_m = -k_S \Delta(t)$$

$$5) \quad \frac{1}{V_1} \frac{dN_1(t)}{dt} = \frac{d\pi_{N_1}(t)}{dt} \quad \frac{1}{V_2} \frac{dN_2(t)}{dt} = \frac{d\pi_{N_2}(t)}{dt}$$

$$\frac{d(\Delta t)}{dt} = \frac{d(\pi_{N_1}(t) - \pi_{N_2}(t))}{dt} = \frac{d\pi_{N_1}(t)}{dt} - \frac{d\pi_{N_2}(t)}{dt}$$

$$\frac{d(\Delta t)}{dt} = \frac{1}{V_1} \frac{dN_1(t)}{dt} - \frac{1}{V_2} \frac{dN_2(t)}{dt}$$

$$\frac{d(\Delta t)}{dt} = \frac{k_S \Delta(t)}{V_1} + \frac{k_S \Delta(t)}{V_2} = k_S \Delta(t) \left( \frac{1}{V_1} + \frac{1}{V_2} \right)$$

$$\boxed{\frac{d(\Delta t)}{dt} - \frac{\Delta t}{\tau} = 0}$$

$$\frac{d(\Delta t)}{\Delta t} = -\frac{dt}{\tau} \Rightarrow \Delta(t) = C e^{-\alpha t}$$

$$\alpha = \frac{1}{\tau}$$

$$\Delta(t=0) = \Delta_0 = C$$

$$\Delta(t) = \Delta_0 e^{-t/\tau}$$

$$t_1 \quad T_9 \quad \Delta(t=t_1) = \frac{\Delta_0}{10} = \Delta_0 e^{-t_1/\tau}$$

$$\Rightarrow e^{t_1/\tau} = 10 \Rightarrow t_1/\tau = \ln 10$$

$$t_1 = \tau \ln 10 = 2,3 \tau$$