

Question de cours :

- Polarisation d'un gaz de molécules polaires sous faible pression.

② 1) • électronique
• atomique
• orientable

$$\alpha = \alpha_e + \alpha_a + \alpha_o.$$

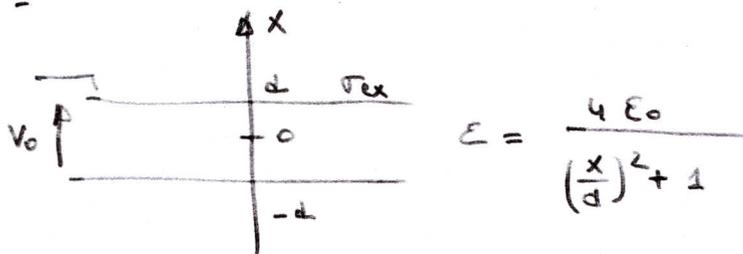
② 2)
$$\vec{P} = N \vec{p} = N \alpha \epsilon_0 \vec{E}_e$$

$$\vec{E}_e = \vec{E} \quad \vec{P} = N (\alpha_e + \alpha_a + \alpha_o) \vec{E} \epsilon_0$$

③ 3)
$$\vec{D} = \epsilon_0 \epsilon_r \vec{E} = \epsilon_0 \vec{E} + \vec{P} = \epsilon_0 (1 + N(\alpha_e + \alpha_a + \alpha_o)) \vec{E}$$

$$\epsilon_r = 1 + N\alpha.$$

Exercice 1 =



① 1) invariances $\forall z$ $\vec{D}(x, z) = \vec{D}(x)$
 ② 2) sym. \forall plan $\parallel x$ = symétrie $\vec{D}(x) = D_x(x) \vec{e}_x$

→ Dans les x discontin.

② 1)
$$\oiint \vec{D} \cdot \vec{n} ds = Q_{ext}.$$



② 2)
$$\oiint \vec{D} \cdot \vec{n} ds = -D_x \cdot \text{circ} = \sigma_{ext} S \quad \text{①} \quad \boxed{D_x = -\sigma_{ext}}$$

$\forall x, -d < x < d.$

① 1)
$$\vec{E} = \frac{\vec{D}}{\epsilon}$$

$$E_x = \frac{D_x \left(\left(\frac{x}{d}\right)^2 + 1 \right)}{4 \epsilon_0 x}$$

*est entre les racines.
 $\vec{E} = -\text{grad } V = -\frac{dV}{dx} \vec{e}_x$*

② 2)
$$V(x) = - \int_{x=d}^x E(x) dx = - \frac{D_x}{4 \epsilon_0} \int_{-d}^x \left(\left(\frac{x}{d}\right)^2 + 1 \right) dx$$

$$= - \frac{D_x}{4 \epsilon_0} \left[\frac{1}{3} \frac{x^3}{d^2} + x \right]_{-d}^x$$

$-\frac{dV}{dx} = -d$

$$V_a - V_b = \frac{Q}{4\pi\sqrt{\epsilon_1\epsilon_2}} \int_{x_a}^{x_b} \frac{1}{1+x^2} dx = \frac{Q}{4\pi\sqrt{\epsilon_1\epsilon_2}} \left[\tan^{-1} x \right]_{x_a}^{x_b}$$

$$V_a - V_b = \frac{Q}{4\pi\sqrt{\epsilon_1\epsilon_2}} \cdot \left(\frac{\epsilon_1}{\epsilon_2} \right)^{1/2} (\tan^{-1} x_a - \tan^{-1} x_b) !$$

$$Q = CV$$

$$C = 4\pi\sqrt{\epsilon_1\epsilon_2} \left(\frac{\epsilon_1}{\epsilon_2} \right)^{1/2} (\tan^{-1} x_a - \tan^{-1} x_b)^{-1}$$

1) [4]